

Cherimonaine, a Novel Dimeric Amide from the Stems of *Annona cherimola*

Chung-Yi Chen, Fang-Rong Chang and Yang-Chang Wu*

Graduate Institute of Natural Products, Kaohsiung Medical College, Kaohsiung, Taiwan, R.O.C.

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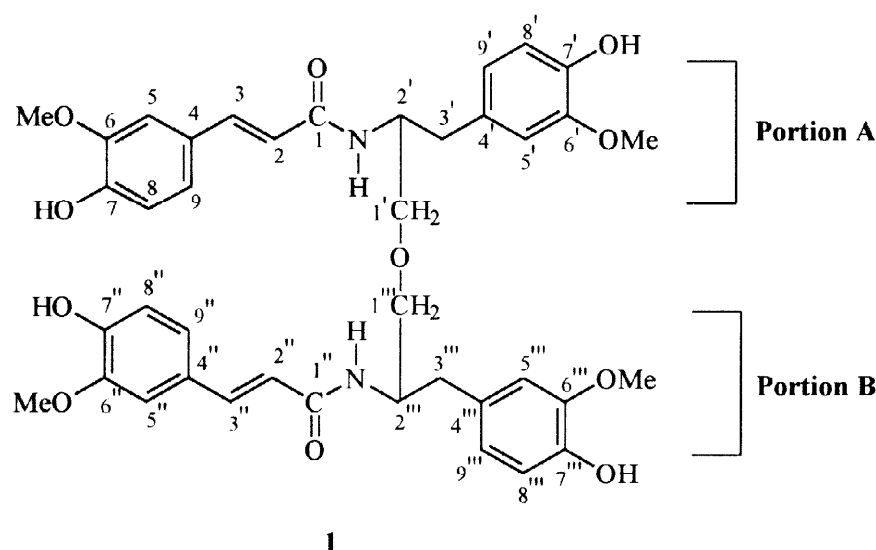
Abstract: Cherimonaine (**1**), a novel *Annona* dimeric amide, has been isolated from *Annona cherimola*, and its structure was determined on the basis of spectroscopic analysis. Amide **1** consists a unique ether bridge between two monomeric amides which are composed from ferulic acid and 6-methoxyl-7-hydroxyl amphetamine. © 1997 Elsevier Science Ltd. All rights reserved.

In the course of screening for biologically and chemically novel agents from Formosan Annonaceous plants, we found that *Annona cherimola* (Annonaceae) producing a novel dimeric amide named cherimonaine (**1**). Previously, we have isolated a novel alkaloid, cherimoline,¹ and twenty-one alkaloids, four kauranes, two amides, one purine, one lactam amide and six steroids from this plant.² In this paper, we report the isolation and structural elucidation of this new compound (**1**).

A number of Annonaceae and *A. cherimola* in particular have been described as cytotoxic.³ They are used in the folk medicine of some tropical countries to treat various tumours and cancers.⁴ The fresh stems of the plant were extracted with MeOH. The extracts was concentrated *in vacuo* and partitioned between CHCl₃ and water. The organic layer was separated by silica gel column chromatography using gradient elution of CHCl₃-MeOH. Further purification by preparative TLC on silica gel with hexane-EtOAc (1:4) afforded a white powder of 10 mg cherimonaine (**1**).

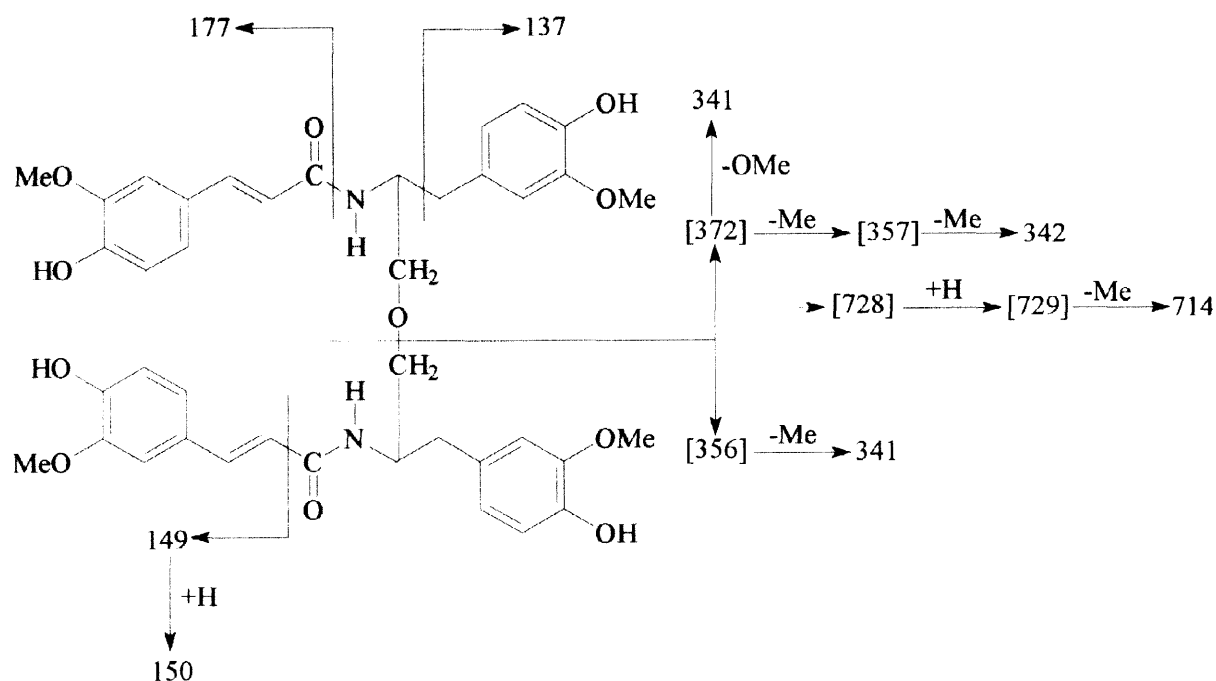
Cherimonaine (**1**), was obtained as white amorphous powder with $[\alpha]_D = \pm 0^\circ$ (c 0.13, MeOH). The UV spectrum with strong absorptions maxima at 220, 290 and 317 nm, and the IR spectrum absorption at λ 1690 cm⁻¹ revealed an α , β -unsaturated amide carbonyl functionality.⁵⁻¹⁰ The structure determination commenced with the molecular formula of C₄₀H₄₄N₂O₁₁ established by HRFABMS at m/z [M+H-CH₃]⁺ 714.2786 (Δ 0.3 mmu). The ¹H NMR spectrum of **1** showed an ABX pattern at δ 7.20 (1H, d, J=8.0), 7.31 (1H, dd, J=8.0, 2.0) and 7.39 (1H, d, J=2.0) for H-8, H-9 and H-5 in the ferulic acid moiety, respectively. A downfield doublet at δ

8.05 ($J=16.0$ Hz) was assigned to be the C-3 olefinic proton showing *trans*-coupling with the C-2 olefinic proton which appeared as a doublet at δ 6.80 ($J=16.0$ Hz). Another ABX pattern at δ 6.92 (1H, dd, $J=8.0, 1.6$), 7.00 (1H, d, $J=1.6$) and 7.20 (1H, d, $J=8.0$) for H-9', H-5' and H-8' in the 6-methoxy-7-hydroxyl amphetamine moiety, respectively. This was assigned as the C-2' methine proton due to the coupling with the neighbor C-3' and C-1' methylene protons which appeared as two double doublet at δ 2.92 ($J=14.0, 7.6$ Hz), δ 3.06 ($J=14.0, 6.8$ Hz) and δ 4.57 ($J=11.4, 5.2$ Hz) and δ 4.80 ($J=11.4, 6.0$ Hz), respectively. The ^{13}C NMR (Table 1) and DEPT experiments of **1** showed twenty resonance lines consisting of two methyls, two methylenes, nine methines, and seven quaternary carbons (including a carbonyl signal at δ 167.2).



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The structure **1** was also confirmed by 2D NMR experiments. A COSY correlation was observed between the H-2' and H-1' , and between H-2' and H-3'. The HETCOR experiment showed that the carbon signals at δ 64.7 for C-1', 40.9 for C-2' and 35.3 for C-3' were correlated to the proton signals at δ 4.80 and 4.57 for H-1', δ 2.66 for H-2' ,and δ 3.06 and 2.92 for H-3', respectively. Thus, these two portions of the dimeric amide were proved to be ether linked at the C-1' position of the A-portion and at the C-1''' position of the B-portion. The NOESY correlations between H-2' and H-3' and between H-2' and H-1' established the connective site as shown in structure **1**. This dimer structure was further confirmed by FABMS fragment at 341 $[\text{M}-\text{C}_{21}\text{H}_{25}\text{O}_6\text{N}]^+$ (Scheme 1). Compound **1** showed zero in optical rotation, suggested **1** as a *meso* compound. Alkaloids, which possess a dimeric structure, are a rare class of compounds except for the indole and the benzyloquinoline alkaloids.¹¹ This is the first example of the dimeric *Annona* acid amide composed of ferulic acid and 6-methoxyl-7-hydroxy amphetamine.



Scheme 1. Ms fragment ions of **1** (m/z)

Table 1. ^{13}C (100 MHz, methanol- d_4) and ^1H NMR (400 MHz, methanol- d_4) data of ferulic acid moiety of cherinonaine (**1**).

C#	δ_{C}	δ_{H}	mult., J (Hz)	COSY	NOESY
1(1'')	167.23				
2(2'')	114.27	6.80	d, 16.0	H-3(3'')	H-3(3'')
3(3'')	144.05	8.05	d, 16.0	H-2(2'')	H-2(2'')
4(4'')	131.59				
5(5'')	109.67	7.39	d, 2.0		6(6'')-OMe
6(6'')	148.45				
7(7'')-OH	147.07	5.01	br s		H-8(8'')
8(8'')	114.94	7.20	d, 8.0	H-9(9'')	H-9(9'')
9(9'')	123.01	7.31	dd, 8.0, 2.0	H-8(8'')	H-8(8''), H-5(5'')
6(6'')-OMe	55.90	3.78			H-5(5'')

Table 2. ^{13}C (100 MHz, methanol- d_4) and ^1H NMR (400 MHz, methanol- d_4) data of 6-methoxy-7-hydroxy amphetamine moiety of cherinonaine (1).

C#	δ_{C}	δ_{H}	mult., J(Hz)	COSY	NOESY
1'(1''')	64.42	1'(1''')a : 4.80 1'(1''')b : 4.57	dd, 11.4, 6.0 dd, 11.4, 5.2	H-2'(2''')	H-1'(1''')b, H-2'(2''')
2'(2''')	40.16	2.66	m	H-1'(1'''), H-3'(3''')	H-3'(3''')a, H-3'(3''')b, H-1'(1''')a
3'(3''')	35.22	3'(3''')a : 3.06 3'(3''')b : 2.92	dd, 14.0, 6.8 dd, 14.0, 7.6	H-2'(2''')	H-3'(3''')b H-3'(3''')a, H-2'(2''')
4'(4''')	126.62				
5'(5''')	111.39	7.00	d, 1.6		6'(6''')-OMe
6'(6''')	146.60				
7'(7''')-OH	145.15	5.01	br s		H-8'(8''')
8'(8''')	115.00	7.20	d, 8.0	H-9'(9''')	H-9'(9''')
9'(9''')	121.67	6.92	dd, 8.0, 1.6	H-8'(8''')	H-8'(8'''), H-5'(5''')
6'(6''')-OMe	55.72	3.76			H-5'(5''')

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References and notes.

- Chen, C. Y., Chang, F. R. and Wu, Y. C. *Tetrahedron Lett.* **1997**, 38, 6247.
- Chen, C. Y., Chang, F. R. and Wu, Y. C. *J. Chin. Chem. Soc.* **1997**, 44, 313.
- McKenna, G. F. and Taylor, A. *Texas Dept. Biol. Med.* **1962**, 20, 314.
- Barriga, H. G., *Flora Medicinal de Colombia*, Bogota', **1974**, vol. 1, p. 340.
- Rahman, A. U., Bhatt, M. K., Akhtar, F. and Choudhary, M. I. *Phytochemistry* **1992**, 31, 2869.
- Wu, Y. C., Chang, G. Y., Ko, F. N. and Teng, C. M. *Planta Med.* **1995**, 61, 146.
- Fukuda, N., Yonemitsu, M. and Kimura, Y. *Chem. Pharm. Bull.* **1983**, 31, 156.
- Heerden, F. R. V., Braudt, F. V. and Roux, D. G. *J. Phytochemistry* **1980**, 19, 2125.
- Chatterjee, K., Dhara, P., Rej, R. N. and Ghosh, P. C. *Phytochemistry* **1977**, 16, 397.
- Uemura, S., Tonakai, S., Yamauchi, T., Nishimura, F., Mizutaki, S. and Tamaki, K. *Chemistry Express* **1987**, 2, 433.
- Ichiro, T., Ichiro, Y., Motohiro, N., Yukio, H., Koichi, T. and Hideji, I. *Tetrahedron Lett.* **1996**, 37, 7053.